ORIGINAL ARTICLE



Penisporolides A and B, Two New Spiral Lactones from the Marine-derived Fungus *Penicillium* sp.

Xiang Li, Isabel Sattler, Wenhan Lin

Received: November 4, 2006 / Accepted: February 17, 2007 © Japan Antibiotics Research Association

Abstract In our continued screening on bioactive constituents from marine-derived fungi, two novel compounds containing a rare spiral-lactone skeleton were isolated from lyophilized culture broth of the marine-derived fungus *Penicillium* sp. The structures of penisporolides A and B were elucidated on the basis of extensive 1D and 2D NMR as well as HRESI-MS spectroscopic data analysis. The relative stereochemistries of the compounds were assessed by analysis of NOESY data together with the comparison with data from previous literatures.

Keywords penisporolide A, penisporolide B, marinederived fungus, *Penicillium* sp.

Introduction

Antimicrobial and cytotoxic activities [1, 2] were discovered from *Penicillium* sp. (HKI Strain No. GT2002605), an organism that was purified from the mangrove plant *Kandelia candel*. From lyophilized culture broth of this fungus, we discovered two novel compounds containing a rare spiral-lactone skeleton, namely penisporolides A (1) and B (2). The structures of the compounds were elucidated on the basis of MS, ¹H and ¹³C NMR, including 2D NMR spectroscopic analysis. This paper describes the structure elucidation of the obtained compounds.

Results and Discussion

Continuing the chemical study of marine-derived fungi, we isolated and characterized two new lactones 1 and 2 from the methanol extract of marine-derived fungus *Penicillium* sp.

The first compound isolated was a pale yellow oil with the molecular formula $C_{18}H_{30}O_5$, as deduced from HRESI-MS m/z: 349.2033 [M+Na]⁺ (calcd 349.1991) and ¹³C NMR spectroscopic data, requiring four sites of unsaturation. The IR spectrum of 1 showed bands characteristic of lactone carbonyl ($v_{max} = 1774 \text{ cm}^{-1}$) functionality. The ¹³C NMR spectrum comprised eighteen carbon signals with the following multiplicities (DEPT spectrum): three quaternary carbons at δ 180.8 (C-1), 44.4 (C-2) and 115.5 (C-6), four oxymethine carbons at δ 85.2 (C-3), 80.5 (C-4), 78.8 (C-9) and 71.5 (C-16), eight methylene carbons at δ 42.4 (C-5), 35.5 (C-7), 30.1 (C-8), 30.0 (C-12), 25.6 (C-13), 18.8 (C-14), 37.8 (C-15) and 39.7 (C-17), three methyl carbons at δ 23.1 (C-10), 18.1 (C-11) and 9.8 (C-18). Comprehensive analysis of 2D NMR data, including the results of ¹H-¹H COSY, HMQC, and HMBC experiments have been used to elucidate the planar structure of 1.

Analysis of the ¹H-¹H COSY (Fig. 1) and HMQC spectra suggested the presence of three ¹H-¹H spin systems: $H_3-H_4-H_5$, $H_7-H_8-H_9$, $H_{12}-H_{13}-H_{14}-H_{15}-H_{16}-H_{17}-H_{-18}$. The long-rang ¹H-¹³C correlations were observed by HMBC spectrum (Fig. 1 and Table 1) as following: from C-1 to H-3, H-4, H-10, H-11; from C-2 to H-4, H-10, H-11; from C-3 to H-4, H-5a, H-10, H-11; from C-5 to H-4, H-7; from C-6 to H-3, H-4, H-5, H-7, H-8, H-9; from C-7 to H-5, H-9; from C-8 to H-5, H-9, H-12; from C-9 to H-7, H-13. The skeleton of the compound was established by comparison the spectral data with those of cephalosporolide E, F and

X. Li (Corresponding author), W. Lin: State Key Laboratory of Natural and Biomimetic Drugs, Peking University, Beijing 100083, P.R. China, E-mail: lixi@agr.gc.ca

X. Li, I. Sattler: Hans-Knöll-Institute for Natural Products Research, Beutenbergstraße 11, D-07745, Jena, Germany

ascospiroketal A, B [5~7]. ¹³C NMR data of 1 with those of former reported compounds showed very similar chemical shifts of C-1, C-4, C-5, C-6, C-7, C-8 and C-9. The carbon signals of C-2 and C-3 shifted upfield to δ 44.4 and 85.2, respectively, due to the dimethylation of 2position. The side chain of 1 was determined by ¹H-¹H COSY (Fig. 1) together with HMQC spectral data analysis. All these data above led us to determine the planar structure of 1 as a spiral lactone substituted with long side

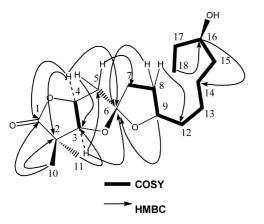


Fig. 1 Main COSY and HMBC correlations of penisporolide A (1).

chain in 9-position.

The relative stereochemistry of 1 was assessed by analysis of NOESY data (Fig. 2) together with the comparison of literatures [7]. The NOESY spectrum showed the correlation between H-3 (δ 4.26) and H-11 (δ 1.24) indicated C-3 had the β -orientation; H-3 showed correlation with H-4 (δ 5. 01) indicated C-4 also had the β orientation; H-5b (δ 2.52) showed cross-peak with H-7 (δ 2.10) and H-3. The similarity of the ¹H and ¹³C NMR spectra of 1 when compared to those of ascospiroketal A [7] indicated that the former natural product possessed the same configuration, but differed in their substituents. This was confirmed by optical rotation comparison with the known compound. 1 possessed optical rotation as $\left[\alpha\right]_{D}^{20}$ +37 (c 0.5 in MeOH) similar with that of ascospiroketal A $([\alpha]_D^{23} + 20 (c \ 0.45 \text{ in MeOH}))$. Thus, we can establish the relative configuration of 1 as showed in Fig. 2.

2 was obtained as pale yellow oil, and its molecular formula was established to be $C_{17}H_{26}O_5$ from HRESI-MS m/z: 333.1682 [M+Na]⁺ (calcd 333.1678). The similarity of the ¹H and ¹³C NMR spectra of **2** (Table 2) when compared to those of **1** (Table 1) indicated that they possessed the same skeleton, but differed in their side chain substituents. The ¹³C NMR spectra of **2** (Table 2) appeared an additional carbonyl group resonating at δ 212.5 (s)

Table 1 ¹H and ¹³C NMR spectral data of penisporolide A in CDCl₃^a

#	¹ H NMR	¹³ C NMR	¹ H- ¹ H COSY	HMBC	NOESY
1	_	180.8 (s)		_	_
2	_	44.4 (s)	_	_	_
3	4.26 (d, J=3.6)	85.2 (d)	H-4	C-1, C-10, C-11, C-5, C-6	H-4, H-5, H-10
4	5.01 (m)	80.5 (d)	H-3, H-5	C-1, C-2, C-3, C-5, C-6	H-3
5a	2.35 (d, <i>J</i> =3.6)	42.4 (t)	H-5b, H-4	C-3, C-6 ,C-7, C-8	_
5b	2.52 (d, <i>J</i> =5.4)		H-5a	C-6, C-7, C-8	H-7, H-3
6	_	115.5 (s)	_	_	—
7	2.10 (m)	35.5 (t)	H-8a, H-8b	C-5 ,C-6, C-9	H-5b
8a	2.15 (m)	30.1 (t)	H-7, H-8b, H-9	C-6, C-12	H-9
8b	2.20 (m)		H-7, H-8a	C-6, C-12	_
9	4.05 (m)	78.8 (d)	H-8a, H-12	C-6, C-7, C-8, C-13	H-8a
10	1.19 (s)	23.1 (q)	_	C-1, C-2, C-3, C-11	H-3
11	1.24 (s)	18.1 (q)	_	C-1, C-2, C-3, C-10	_
12	1.55 (m)	30.0 (t)	H-9, H-13	C-8, C-14	_
13	1.50 (m)	25.6 (t)	H-12, H-14	C-9, C-12, C-15	_
14	1.53 (m)	18.8 (t)	H-13, H-15	C-12, C-16	_
15	1.50 (m)	37.8 (t)	H-14, H-16	C-13, C-16, C-17	_
16	3.50 (m)	71.5 (d)	H-15, H-17	C-14, C-18	_
17	1.50 (m)	39.7 (t)	H-16, H-18	C-15	_
18	0.90 (t, J=4.0)	9.8 (q)	H-17	C-16, C-17	_

° Chemical shift δ and (multiplicity, J in Hz). — No signal appeared in spectra.

which showed HMBC correlation to δ 2.43 (2H, t, J=4.2, H-13) and 2.46 (2H, t, J=4.2, H-15) indicated that C-14 had been substituted with ketone group. This observation together with NMR data comparison with 1 led us to establish the structure of **2**. Compounds **1** and **2** showed no effects on xanthine oxidase, other bioactivities are under investigation.

Experimental

General

¹H and ¹³C NMR spectra were measured on a Bruker Avance DRX 500 spectrometer using TMS as an internal standard. Chemical shifts (δ) expressed in parts per million (ppm) and coupling constants (*J*) are reported in Hertz (Hz). ESI-MS spectra were recorded on a triple quadrupole

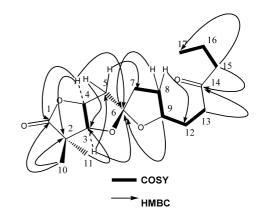


Fig. 3 Main COSY and HMBC correlations of penisporolide B (2).

Fig. 2 Key NOESY correlations of **1**.

Table 2 ¹H and ¹³C NMR spectral data of penisporolide B in CDCl₃^a

¹H NMR ¹³C NMR ¹H-¹H COSY HMBC NOESY 1 180.5 (ND) 2 44.7 (s) 3 4.30 (d, J=3.6) 85.2 (d) H-4 C-1, C-10, C-11, C-5, C-6 H-4, H-5, H-10 4 5.05 (td, J=5.4, 3.6) C-1, C-3, C-5, C-6 80.5 (d) H-3, H-5 H-3 C-3, C-6, C-7, C-8 5a 2.33 (d, J=3.6) 42.5 (t) H-5b, H-4 H-3 2.50 (d, J=5.4) C-6 ,C-7, C-8 5b H-5a H-7 115.5 (ND) 6 7 2.09 (m) 35.5 (t) H-8a, H-8b C-5, C-6, C-9 H-5b H-7, H-8b, H-9 C-6, C-12 8a 2.16~2.18 (m) 30.0 (t) H-9 C-6, C-12 8b 2.16~2.18 (m) _____ H-7, H-8a H-8a, H-12 C-7, C-8, C-13 H-8a 9 4.05 (t, J=3.6) 78.5 (d) 10 1.23 (s) 23.1 (q) C-1, C-3, C-11 H-3 ____ 11 1.29 (s) 18.1 (q) C-1, C-3, C-10 H-9, H-13 C-8, C-14 12 1.50 (m) 20.1 (t) C-12, C-14 13 2.43 (t, J=4.2) 34.9 (t) H-12 14 212.5 (ND) 15 2.46 (t, J=4.2) 42.3 (t) H-16 C-14, C-17 16 1.27 (m) 17.3 (t) H-15, H-17 C-14, C-17 C-15 17 0.95 (t, J=6.9) 13.7 (q) H-16

^a Chemical shift δ and (multiplicity, *J* in Hz). ND, The signal was not detected in survey spectra, but was apparent in HMBC spectra. — No signal appeared in spectra. mass spectrometer Quattro (VG Biotech, Altrincham, England) and HRESI-MS spectra were obtained on a Bruker FT-ICRMS spectrometer. Column chromatography was carried out with Silica gel (Merck GMBH Germany, 200~300 mesh), Lichrospher RP-8 (20 μ m), TLC: silica gel plates (Macherey-Nagel, SilG/UV254, 0.20 mm), spots were detect by UV₂₅₄ and anisaldehyde/H₂SO₄ (10%); Sephadex LH-20 (Pharmacia); Fermentation: 200 liters fermenter (Braun Diessel).

Penicillium Strain

The strain (HKI strain GT2002605) was purified from mangrove plant *Kandelia candel* that was collected from Hainan Island, Southern China, in October 2002. The species was identified by Prof. Peng Lin from Xiamen University. The strain was stored in Hans-Knöll-Institute of Natural Products Research Jena, Germany. The strain was cultured at 21°C in 300 liter fermenter with control of pH as 5.5 for 7 days.

Culture Conditions

The spores of strain growing on agar slants (artificial sea salt solution 800 ml/liter, nalidixic acid 20 mg/liter and cycloheximide 30 mg/liter) was transferred to a flask (20 ml) and was inoculated on liquid medium (glucose 5 g/liter, peptone 1 g/liter, yeast extract 0.5 g/liter, beef extract 0.5 g/liter, NaCl 3 g/liter). The flask was incubated at 28°C on a rotary shaker for 48 hours, and the mycelium was transferred to a 1-liter Erlenmeyer flask containing culture liquid (400 ml). After 48 hours, the mycelium was transferred to 200 liter fermenter then incubated at 28°C with pH controlled at 5.5 for 7 days.

Extraction and Isolation

The lyophilized culture broth was extracted with MeOH at room temperature. The extract was concentrated under reduced pressure to give the dark brown oily residue (27 g) that was fractionated by reverse phase (C-8) using H₂O, aqueous MeOH (30%, 60%, 90%) and MeOH to give four fractions: the 90% MeOH fraction was further fractionated on a Silica gel column (*n*-hexane : EtOAc=20 : 1, 5 : 1, 1 : 1) to get four fractions, the third part of fraction one was separated by column chromatography with Sephadex LH-20 (CHCl₃ : MeOH=1 : 1) to purify **1** (86 mg, yield: 3.0×10^{-3}) and **2** (4.2 mg, yield: 0.15×10^{-3}).

Penisporolide A (1)

1 was obtained as a pale yellow oil; $[\alpha]_D^{20} + 37$ (*c* 0.5 in MeOH). UV λ_{max} (MeOH) nm: 202, 224 (sh). IR v_{max}^{KBr} cm⁻¹: 2951, 2936, 1774. ESI-MS [+]: *m*/*z*=675.36 [2M+Na]⁺, 349.22 [M+Na]⁺, 327.22 [M+H]⁺; HRESI-MS *m*/*z*:

349.2033 [M+Na]⁺ (calcd for $C_{18}H_{30}O_5Na$, 349.1991); ¹H and ¹³C NMR data see Table 1.

Penisporolide B (2)

2 was obtained as a pale yellow oil; $[\alpha]_D^{20} + 81$ (*c* 1.0 in MeOH). IR $v_{\text{max}}^{\text{KBr}}$ cm⁻¹: 2957, 2935, 1765, 1743. ESI-MS [+]: m/z=311.77 [M+H]⁺; HRESI-MS m/z: 333.1682 [M+Na]⁺ (calcd for C₁₇H₂₆O₅Na, 333.1678). ¹H and ¹³C NMR data see Table 2.

Assay for Inhibition of XOD Activity

Xanthine oxidase activity was evaluated by the spectrophotometric measurement of the formation of uric acid from xanthine [8]. A 100 μ M solution of xanthine in 0.1 M phosphate buffer pH 7.8 with 0.05 units/ml of xanthine oxidase was incubated for 15 minutes at room temperature and read at 295 nm against a control sample, which did not contain the enzyme. Five gradient concentrations of compound **1** and **2** (10, 50, 100, 200, 400 μ g/ml) were added to samples before the enzyme had been added, and their effect on the generation of uric acid was used to calculate the IC₅₀ value.

Acknowledgments We are grateful to Dr. F. Gollmick (HKI) and Professor Zhiwei Deng (Beijing Normal University) for acquiring the NMR spectra. This project is part of a collaboration between the Hans-Knöll-Institute for Natural Product Research (HKI) and the National Research Laboratory of Natural and Biomimetic Drugs, Peking University (Beijing) and was funded by the Germany Federal Ministry for Education and Research and Technology (BMBF 0312849A, CHN02/323 and CHN 02/322).

References

- Blunt JW, Copp BR, Munro MHG. Marine natural products. Nat Prod Rep 22: 1–61 (2005)
- Grabley S, Thiericke R. *In* Drug Discovery from Nature. Springer Verlag, Berlin, pp. 124–125 (1999)
- Afghan F, John G, James R, Jacqueline AT. Two C₁₀ lactones from *Cephalosporium Aphidicola*. Phytochemistry 38: 557–558 (1995)
- Vatcharin R, Songyos P, Chaveng P, Masahiko I, Sumilee S. 10-Membered macrolides from the insect pathogenic fungus *Cordyceps militaris* BCC 2816. J Nat Prod 67: 1953–1955 (2004)
- Ackland MJ, Hanson JR, Hitchcock PB, Mabelis RP, Ratcliffe AH. The structure of thiobiscephalosporolide A, a dimeric pentaketide macrolide from *Cephalosporium aphidicola*. J Chem Soc Perkin Trans I 5: 2755–2757 (1984)
- Ackland MJ, Hanson JR, Hitchcock PB, Ratcliffe AH. Structures of the cephalosporolides B~F, a group of C₁₀

lactones from *Cephalosporium aphidicola*. J Chem Soc Perkin Trans I 2: 843–847 (1985)

7. Seibert SF, Krick A, Eguereva E, Kehraus S, Konig GM. Ascospiroketals A and B, unprecedented cycloethers from

the marine-derived fungus *Ascochyta salicorniae*. Organic Lett 9: 239–242 (2007)

8. Robak J, Gryglewsky RJ. Flavonoids are scavengers of superoxide anions. Biochem J 37: 837–841 (1988)